THE CONVERSION OF SHIONONE INTO DIHYDROBACCHARIS OXIDE

Kazuo Tachibana and Takeyoshi Takahashi Department of Chemistry, Faculty of Science, The University of Tokyo, Bunkyo-ku, Tokyo, Japan

(Received in Japan 24 March 1975; received in UK for publication 22 April 1975)

The structure of baccharis oxide, a triterpene oxide isolated from <u>Baccharis</u> <u>halimifolia</u> L.,^{1b)} has been determined to be <u>1</u> by Norwegian workers.^{1a)} A close biogenetic relationship between baccharis oxide (<u>1</u>) and shionone (<u>2</u>)^{2a)} led us to undertake a study to correlate these two triterpenes each other. We wish to report the conversion of shionone (<u>2</u>) into dihydrobaccharis oxide (<u>3</u>),^{1b)} a hydrogenated product of baccharis oxide (<u>1</u>).

Shionone (2) was transformed <u>via</u> shionan-3-one, 2b,c shionan-3 β -ol, 2b,c and shion-3-ene, 2c into 3 β ,4 β -epoxyshionane (4)³) by the known procedures. Treatment of the β -epoxide (4; m.p. 154-155°, 313 mg) in ether with BF₃-Et₂O at -15° for 30 min gave a mixture showing five spots on silica gel TLC impregnated with silver nitrate; a complete disappearance of the starting material was observed. The material corresponding to the least polar spot was separated by silica gel column chromatography to afford an oxide (2; 44 mg), m.p. 127-128.5°, [α]_D+45° (CHCl₃; c 1.7), M⁺ at m/e 428 (C₃₀H₅₂O); PMR (CDCl₃): δ 3.73 (d, J = 5 Hz; C₍₃₎-H), which was found to be identical (m.p., mixed m.p., [α]_D, IR, PMR, mass spectrum and TLC) with dihydrobaccharis oxide (2).^{1a,b}) One of the other four products was identified as bacchar-12-en-3 β -ol (5).^{1b,3}) Identification of the other three products and examination of reaction conditions (temperature, reaction time, etc.) to form various products mentioned above are under way.

Treatment of the β -epoxide (<u>4</u>) in benzene with the same reagent at 5° (15 min) resulted in the formation of bacchar-12-en-3 β -ol (<u>5</u>) (along with the two among three unidentified products described above); no dihydrobaccharis oxide (<u>3</u>) was obtained. This can be well interpreted by the observation of Norwegian workers^{1b)} that the ether bridge of dihydrobaccharis oxide (<u>3</u>) was easily cleft in similar conditions (BF₃-Et₂0; PhH; room temperature; <u>3</u> min) to give bacchar-12-en-3 β -ol (<u>5</u>).

In connection with the total synthesis of shionone (2) reported recently,⁴⁾ the present communication constitutes formally the total synthesis of dihydrobaccharis oxide ($\underline{3}$).

<u>Acknowledgement</u> : We wish to thank Dr. T. Brunn, Norges Tekniske Høgskole, Trondheim, Norway, for a generous gift of an authentic sample of dihydrobaccharis oxide.

1857



 $\underline{1} \quad R = CH_2CH_2CH=C(CH_3)_2$ $\underline{3} \quad R = CH_2CH_2CH_2CH(CH_3)_2$







REFERENCES

- a- F. Mo, T. Anthonsen, and T. Bruun, <u>Acta Chem. Scand.</u>, <u>26</u>, 1287 (1972).
 b- T. Anthonsen, T. Bruun, E. Hemmer, D. Holme, A. Lamvik, E. Sunde, and N. A. Sørensen, <u>ibid.</u>, <u>24</u>, 2479 (1970). c- E. Suokas and T. Hase, <u>ibid.</u>, <u>25</u>, 2359 (1971).
- a- Y. Moriyama, Y. Tanahashi, T. Takahashi, and G. Ourisson, <u>Bull. Soc. Chim.</u> <u>France, 1968</u>, 2890; T. Tsuyuki, T. Hoshino, M. Ito, and T. Takahashi, <u>ibid.</u>, <u>1968</u>, 2895. And the references cited therein. b- M. Takahashi, W. Kamisako, S. Ishimasa, and K. Miyamura, <u>Yakugaku Zasshi</u>, <u>79</u>, 1281 (1959); W. Kamisako and M. Takahashi, <u>ibid.</u>, <u>84</u>, 318 (1964). c- Y. Tanahashi, T. Takahashi, F. Patil, and G. Ourisson, <u>Bull. Soc. Chim. France</u>, <u>1964</u>, 584.
- S. Yamada, S. Yamada, Y. Moriyama, Y. Tanahashi, and T. Takahashi, <u>Tetrahedron Lett.</u>, <u>1972</u>, 5043.
- 4) R. E. Ireland, C. A. Lipinski, C. J. Kowalski, J. W. Tilley, and D. M. Walba, J. Amer. Chem. Soc., <u>96</u>, 3333 (1974).